## Reactions of μ-η<sup>2</sup>-Allenyl Complexes: Coupling of Diphenylacetylene with Ru<sub>2</sub>(CO)<sub>6</sub>-[u-o,n<sup>2</sup>-PhCC=CH<sub>2</sub>](u-PPh<sub>2</sub>) and the X-Ray Crystal Structure of a Novel Tetranuclear  $Cyclopentadienyl Complex [Ru<sub>2</sub>(CO)<sub>4</sub>{ $\mu$ - $\sigma$ (C,O), $\eta$ <sup>7</sup>-C<sub>5</sub>MePh<sub>2</sub>(C<sub>6</sub>H<sub>4</sub>)(O){ $\mu$ -PPh<sub>2</sub>]}<sub>2</sub>·2C<sub>7</sub>H<sub>8</sub>$

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The  $\mu$ - $\sigma$ , $\eta$ <sup>2</sup>-allenyl complex  $Ru_2(CO)_{6}[\mu$ - $\sigma$ , $\eta$ <sup>2</sup>-PhCC=CH<sub>2</sub> $](\mu$ -PPh<sub>2</sub>) reacts with diphenylacetylene to afford a novel tetranuclear cyclopentadienyl complex  $[Ru_2(CO)_4\{\mu \cdot \sigma(C, O), \eta^7-C_5MePh_2(C_6H_4)(O)\}\{\mu \cdot PPh_2\}]_2$  *via* allenyl-alkyne coupling: the structure has been determined by  $X$ -ray diffraction.

The chemistry of binuclear complexes containing  $\mu$ - $\sigma$ , $\eta$ <sup>2</sup>allenyl groups  $R^1C=C=CR^2R^3$  is of interest for several reasons: (i) these hydrocarbyls have cumulated double bonds, one of which is co-ordinated while the other is free when the ligand functions as a three-electron donor;<sup>1</sup> (ii) their isomeric counterparts, the  $\mu$ - $\eta$ <sup>3</sup>-allyl { $R'CC(R^2)C(R^3)$ } compounds are well known and synthetically useful; $2,3$  and (iii) the parent allenyl ligand  $-CH=C=CH_2$  is composed of three  $C_1$  fragments, carbide (C), methylidyne (CH), and methylene (CH<sub>2</sub>), which have been implicated in various catalytic processes.4 However, the chemistry of these binuclear allenyl compounds is relatively unexplored.<sup>2</sup> We have previously<sup>1</sup> synthesised a series of binuclear allenyl complexes  $[M_2(CO)_6(\mu-\sigma,\eta^2-$ PhC=C=CRR'}( $\mu$ -PPh<sub>2</sub>)] ( $\dot{M}$  = Fe, Ru, or Os; R = R' = H, Me, or Ph) *via* the addition of diazoalkanes to acetylides  $[M_2(CO)_6(\mu-\sigma,n^2-C\equiv CPh)(\mu-PPh_2)]$  and we describe here an unusual coupling reaction with diphenylacetylene which affords a novel tetranuclear cyclopentadienyl complex.

Treatment of the allenyl complex **(1)** (0.25 g, 0.37 mmol) with  $PhC_2Ph$  (0.10 g, 0.56 mmol) in refluxing toluene (35 ml) for 2 h followed by column chromatography [Florisil; eluant  $n-C_7H_{16}-C_7H_8$  (5:1)] afforded a yellow band which gave pale yellow crystals of the cyclopentadienyl complex **(2)** (57%) (from  $CH_2Cl_2-C_7H_8$  at  $-5^{\circ}C$ ) (Scheme 1). The only other band (red-brown) afforded, as an orange powder (11%), an as yet uncharacterised complex (from n-heptane). Spectroscopic data for (2) were as follows: i.r.  $(C_6H_{12})$  v(CO) 2098vs, 2048s, 2023vs, 1997s, and 1974s cm<sup>-1</sup>; <sup>31</sup>P-{<sup>1</sup>H} n.m.r. (CDCl<sub>3</sub>;





**Figure 1.** The molecular structure of  $[Ru_2(CO)_4(\mu\text{-}o(C,O),\eta\text{-}C_5MePh_2(C_6H_4)(O)](\mu\text{-}PPh_2)]_2\text{-}2C_7H_8$  (2) showing the atomic numbering. Important bond distances (Å) and angles (°) not included in the text are:  $Ru(1) - P2.409(2)$ ,  $Ru(2) - P2.346(2)$ ,  $Ru(2) - O(5)$  2.119(3),  $Ru(2) - O(5)$ 2.229(3), Ru(1)-C(5) 2.288(5), Ru(1)-C(6) 2.233(5), Ru(1)-C(7) 2.231(5), Ru(1)-C(8) 2.270(5), Ru(1)-C(9) 2.277(5), C(5)-C(6) 1.439(7), C(5)-O(5) 1.318(6), C(6)-C(7) 1.441(7), C(7)-C(8) 1.431(8), C(9)-C(5) 1.428(7), C(8)-C(9) 1.450(8); Ru(2)-O(5)-Ru(2') 105.0(1),<br>Ru(2)-O(5)-C(5) 115.0(2), Ru(2')-O(5)-C(5) 128.1(2), C(4)-Ru(2)-C(12) 175.7(5), C(3)-Ru(2)-O(5  $Ru(2)-O(5)-C(5)$  115.0(2),  $Ru(2')-O(5)-C(5)$  128.1(2),  $C(4)-Ru(2)-C(12)$  175.7(5),  $C(3)-Ru(2)-O(5)$  179.0(2),  $P-Ru(2)-O(5)$  89.9(1),  $P-Ru(2)-O(5')$  163.5(1).

101.26 MHz; 298 K) **8** +52.5; IH n.m.r. (CDC13; 250.13 MHz; 298 K)  $\delta$  2.4 (d, 6 H, J<sub>PH</sub> 1.5 Hz, CH<sub>3</sub>) and 7.0–8.0 (m, 48 H,  $C_6H_5$  and  $C_6H_4$ ); <sup>13</sup>C-{<sup>1</sup>H} n.m.r. (CDCl<sub>3</sub>; 62.87 MHz; for numbering see Figure 1)  $\delta$  201.6 (d, CO,  $^{2}J_{PC}$  7.6 Hz), 198.5 (d, CO,  $^{2}J_{PC}$  6.7 Hz), 196.2 (d, CO,  $^{2}J_{PC}$  9.6 Hz), 185.0 (d, CO,  $2J_{\text{PC}}$  5.2 Hz), 170.2 [d, C(5,5'),  $3J_{\text{PC}}$  5.6 Hz], 161.9 [d,  $C(12,12^{7})$ ,  $^{2}J_{PC}$  10.9 Hz], 145.9 (d, P–C<sub>i</sub>,  $^{1}J_{PC}$  22.0 Hz), 144.3  $[s, C(23,23')]$ , 141.9 (d, P–C<sub>i</sub>, <sup>1</sup>J<sub>PC</sub> 27.5 Hz), 133.7 [d, C(ll,ll'), **3Jpc** 3.6 Hz], 132.9 **[s,** C(l4,16,14',16')], 132.1 (d, P-C,, 'Jpc 11.3 Hz), 132.0 *[s,* C(l7,17')], 131.9 **[s,** C(13,13'], 131.1 (d, P-C,, 2Jpc 9.4 Hz), 129.8 **[s,** C(24,28,24', 28')], 128.5 (d, P-C<sub>p</sub>,  $\frac{4J_{\text{PC}}}{2.5\text{ Hz}}$ ), 128.4 [s, C(18,22,18',22')], 128.3 (d, P–C<sub>p</sub>, <sup>4</sup>J<sub>PC</sub> 2.4 Hz), 128.1 (d, P–C<sub>m</sub>, <sup>3</sup>J<sub>PC</sub> 10.7 Hz), 128.0 (d, P-Cm, 3Jpc 9.8 Hz), 127.9 *[s,* C(15,15')], 127.7 *[s,*  C(25,27,25',27')], 126.5 **[s,** C(l9,21,19',21')], 125.2 [s, C(26,26')], 124.4 [s, C(20,20')], 101.9 [s, C(8,8')], 98.3 [d,  $C(7,7')$ ,  $5J_{PC}$  2.4 Hz], and 15.1 [s,  $C(10,10')$ ]. These suggested the presence of four terminal carbonyl groups, an opened phosphido bridge, and a new ligand system with two unique deshielded carbon atoms. Full details of the molecular structure (Figure 1) were revealed by a single-crystal  $X$ -ray diffraction study.<sup>†</sup> The molecule lies on a two-fold crystallo- $C(9,9')$ ,  $4J_{PC}$  4.7 Hz], 91.3 [d,  $C(6,6')$ ,  $4J_{PC}$  2.5 Hz], 86.1 [d,

graphic axis passing through the centre of the  $Ru(2)$ ,  $Ru(2')$ , *0(5),0(5')* ring. None of the metal-metal distances [Ru( 1)- ---Ru(2) 3.8312(6), Ru(2)---Ru(2') 3.4501(7) A] are bonding, and a remarkable feature is that the two halves of the molecule are held together by only the two bridging oxygen atoms *O(5)* and *O(5').* We are unaware of any precedent for such an open chain  $Ru_4$  structure in organoruthenium chemistry. Within each of the binuclear fragments, the two ruthenium atoms are bridged by an open phosphido group,  $Ru(1) - P-Ru(2)$  107.36(3)<sup>o</sup>, and by a new organometallic moiety best described as a deprotonated and ortho-metallated **l-hydroxy-2,4,5-triphenyl-3-methylcyclopentadienyl** ligand. The five membered  $C(5)$ — $C(9)$  ring system is  $\eta^5$ -bound to Ru(l), with the C(6)-phenyl ring bound to Ru(2) *via* a o-bond to the *ortho* position  $C(12)$  [Ru(2)– $C(12)$  2.115(6) Å]. The oxygen atom *O(5)* acts as a three-electron donor to the two symmetry-related ruthenium atoms  $Ru(2)$  and  $Ru(2')$  such that all four metal atoms obey the eighteen-electron rule. The stereochemistry at  $Ru(1)$  and  $Ru(1')$  is typical for an  $\eta^5$ -C<sub>5</sub>R<sub>5</sub>Ru(CO)<sub>2</sub>X fragment; Ru(2) and Ru(2') are octahedral. The new cyclopentadienyl ligand is constituted from a molecule of diphenylacetylene, a CO group, and the original allenyl ligand (PhC=C=CH<sub>2</sub>) of  $(1)$  which provides two ring carbon atoms and two substituents for the five-membered ring. Although the mechanism of formation of **(2)** has not been established, substituent connectivities indicate addition of alkyne and CO to the  $\eta^2$ -co-ordinated double bond of the allenyl ligand. Furthermore, since complexes  $n<sup>5</sup>-C<sub>5</sub>H<sub>5</sub>Ru (CO)(PR<sub>3</sub>)(R)$  are known to undergo metallation readily,<sup>5</sup> it is tempting to suggest that *ortho*-metallation of the phenyl ring of the allenyl fragment occurs subsequent to ring formation. The overall reaction to form **(2)** is reminiscent of the coupling of a Fischer-type carbene with CO and alkynes leading to 4-methoxy-l-naphthols, which has been developed into a useful synthetic strategy by Dotz.6 The formation of the cyclopentadienyl complex **(2)** in good yield from **(1)** and

 $t$  *Crystal data:*  $C_{80}H_{54}Ru_{4}P_{2}O_{10} \cdot 2C_{7}H_{8}$ ,  $M = 1825.83$ , monoclinic, space group  $C_2/c$ ,  $a = 14.123(2)$ ,  $b = 20.352(4)$ ,  $c = 28.070(4)$  Å,  $\beta =$  $98.88(\tilde{1})^{\circ}$ ,  $U = 7971(2)$   $\AA^3$ ,  $Z = 4$ ,  $D_c = 1.521$  g cm<sup>-3</sup>,  $\mu(\text{Mo-}K_{\alpha}) =$ 10.31 cm<sup>-1</sup>,  $F(000) = 3680$ . Diffraction data were collected with a Syntex P<sub>2</sub>, diffractometer ( $\omega$  scans; 294  $\pm$  1 K). The structure solution (Patterson, Fourier methods) and refinement (full-matrix leastsquares; all non-hydrogen atoms anisotropic) were based on 4246 observed intensities  $(I \ge 3\sigma(I))$  from 7032 measured data (2 $\theta \le 50^{\circ}$ ). Final *R* and  $R_w$  values were 0.036 and 0.040. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

 $PhC<sub>2</sub>Ph$  contrasts sharply not only with the sporadic appearance of cyclopentadienyl compounds as trace products in metal carbonyl-alkyne reactions7 but also with the coupling pathways observed for acetylenes with binuclear  $\eta^2$ -bound alkynes<sup>2,8</sup> and  $\sigma$ , $\eta$ -acetylides.<sup>9</sup> A significantly different chemistry for  $\sigma$ , $\eta$ -allenyl species can thus be predicted.

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## **References**

- 1 D. Nucciarone, N. J. Taylor, and **A.** J. Carty, *Organometallics,*  1986, *5,* 1179.
- 2 J. Holton, M. F. Lappert, R. Pearce, and P. **I.** W. Yarrow, *Chem. Rev.,* 1983, **83,** 135.
- 3 (a) **A.** Kuhn and H. Werner, J. *Organomet. Chem.,* 1979,179,421; (b) H. Werner, *Adv. Organomet. Chem.,* 1981, 19, 155.
- 4 See, for example, (a) D. L. Davies, J. C. Jeffrey, D. Miguel, P. Sherwood, and F. G. **A.** Stone, J. *Chem.* Soc., 1987, 454, and references therein; (b) **S. A.** R. Knox, *Pure Appl. Chem.,* 1984,56, 81; (c) E. L. Muetterties and J. Stein, *Chem. Rev.,* 1979, 79, 479; (d) W. **A.** Herrmann, *Adv. Organomet. Chem.,* 1982, **20,** 160; (e) J. E. Bradley, *ibid.,* 1983, **22,** 1; (f) R. *C.* Brady, **111,** and R. Pettit, J. *Am. Chem. SOC.,* 1981, 103, 1287.
- 5 M. I. Bruce, R. C. F. Gardner, and F. G. **A.** Stone, J. *Organomet. Chem.,* 1972, **40,** C39.
- 6 K. H. Dotz in 'Transition Metal Carbene Complexes,' Verlag Chemie, Weinheim, 1983, p. 209.
- 7 (a) W. Hubel in 'Organic Synthesis *via* Metal Carbonyls,' eds. I. Wender and P. Pino, Wiley Interscience, vol. **I,** 1968; (b) E. Sappa, **A.** Tiripicchio, and P. Braunstein, *Chem. Rev.,* 1983, **83,** 203.
- 8 (a) M. Green, P. **A.** Kale, and R. J. Mercer, J. *Chem. SOC., Chem. Commun.,* 1987,375; (b) S. **A.** R. Knox, R. F. D. Stansfield, F. G. **A.** Stone, M. **J.** Winter, and P. Woodward, J. *Chem. SOC., Dalton Trans.,* 1982, 173; (c) R. S. Dickson, G. D. Fallon, F. I. McLure, and R. J. Nesbit, *Organometallics,* 1987, 6, 215.
- 9 W. F. Smith, N. J. Taylor, and **A.** J. Carty, J. *Chem.* Soc., *Chem. Commun.,* 1976, 896.